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# Di- $\mu$ -iodo-bis[(1,10-phenanthroline- $\kappa^2 N, N'$ )copper(I)] acetonitrile solvate

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## metal-organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 295 KMean  $\sigma(C-C) = 0.008 \text{ Å}$ R factor = 0.041wR factor = 0.107 Data-to-parameter ratio = 18.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

## Di- $\mu$ -iodo-bis[(1,10-phenanthroline- $\kappa^2 N, N'$ )copper(I)] acetonitrile solvate

In the title complex,  $[Cu_2I_2(C_{12}H_8N_2)_2]\cdot C_2H_3N$ , two  $Cu^I$  atoms are bridged by two I atoms and each Cu<sup>I</sup> atom is chelated by a 1,10-phenanthroline ligand to give four-coordinate tetrahedral geometry.

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#### Comment

The study of copper(I) complexes with the 1,10-phenanthroline (phen) ligand has attracted considerable attention (Healy et al., 1985; Yu et al., 2001). In our study of the preparation of copper(I) complexes, a binuclear compound, (I), was obtained and structurally characterized.

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The crystallographic analysis reveals that (I) is a binuclear complex, in which each Cu<sup>I</sup> atom is in an N<sub>2</sub>I<sub>2</sub> four-coordinate environment with distorted tetrahedral geometry (Fig. 1). The structure is very similar to that of  $[Cu_2(\mu-I)_2(phen)_2]\cdot MeOH$ (Healy et al., 1985). The Cu−I, Cu−N and Cu···Cu distances in both complexes are similar. In (I), the dihedral angle between the two Cu<sub>2</sub>I planes is 16.74 (2)° and the two phen ligands are almost parallel, with a dihedral angle of 1.80 (17)°. Interestingly, the  $[Cu_2(\mu-I)_2(phen)_2]$  molecules form a simple

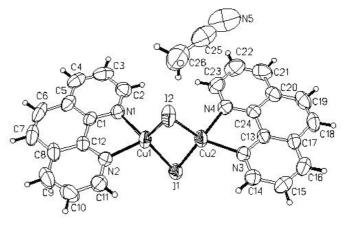


Figure 1 ORTEPII plot (Johnson, 1976) of (I), with displacement ellipsoids drawn at the 50% probability level and H atoms drawn as spheres of arbitrary

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one-dimensional chain *via* face-to-face  $\pi$ - $\pi$  interactions between adjacent phen ligands. The distance between the parallel phen planes is 3.46 (4) Å.

#### **Experimental**

A mixture of CuI (0.019 g, 0.1 mmol) and 1,10-phenanthroline (0.018 g, 0.1 mmol) in CH<sub>3</sub>CN (8 ml) in a 15 ml stainless steel reactor with a Teflon liner was heated to 453 K for 24 h. The reactor was cooled to room temperature at a rate of 10 K h<sup>-1</sup>. Red crystals of the title complex were collected in 57% yield. Elemental analysis calculated for  $C_{26}H_{19}Cu_2I_2N_5$ : C 39.99, H 2.42, N 8.97%; found: C 39.9 5, H 2.44, N 8.96%. IR (KBr, cm<sup>-1</sup>): 3054 (w), 2243 (w), 1957 (w), 1617 (m), 1581 (m), 1507 (m), 1421 (s), 1216 (w), 1152 (w), 1088 (w), 1053 (w), 988 (w), 837 (s), 713 (m), 735 (s), 627 (m), 417 (w).

#### Crystal data

$[Cu_2I_2(C_{12}H_8N_2)_2]\cdot C_2H_3N$	$D_{\rm r} = 1.972 \; {\rm Mg \; m^{-3}}$
$M_r = 782.36$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 4519
a = 10.4920 (6)  Å	reflections
b = 23.6166 (13)  Å	$\theta = 2.6 - 26.0^{\circ}$
c = 10.6391 (6) Å	$\mu = 3.98 \text{ mm}^{-1}$
$\beta = 91.436 \ (1)^{\circ}$	T = 295 (2)  K
$V = 2635.4 (3) \text{ Å}^3$	Prism, red
Z = 4	$0.25 \times 0.16 \times 0.15 \text{ mm}$

#### Data collection

Bruker SMART APEX area-	5991 independent reflections
detector diffractometer	4466 reflections with $I > 2\sigma(I)$
$\varphi$ and $\omega$ scans	$R_{\rm int} = 0.029$
Absorption correction: multi-scan	$\theta_{ m max} = 27.5^{\circ}$
(SADABS; Sheldrick, 2002)	$h = -13 \rightarrow 13$
$T_{\min} = 0.409, T_{\max} = 0.550$	$k = -30 \rightarrow 30$
16 362 measured reflections	$l = -10 \rightarrow 13$

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0544P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.041$	+ 0.5635P
$wR(F^2) = 0.107$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.03	$(\Delta/\sigma)_{\text{max}} = 0.001$
5991 reflections	$\Delta \rho_{\text{max}} = 1.57 \text{ e Å}^{-3}$
317 parameters	$\Delta \rho_{\min} = -1.06 \text{ e Å}^{-3}$
H-atom parameters constrained	

**Table 1** Selected geometric parameters (Å, °).

I1-Cu1	2.5909 (6)	Cu1-N1	2.078 (3)
I1-Cu2	2.6297 (7)	Cu1-N2	2.085 (3)
I2-Cu1	2.5876 (7)	Cu2-N3	2.099 (4)
I2-Cu2	2.5953 (7)	Cu2-N4	2.080 (4)
	/->		
I1-Cu1-I2	118.75 (2)	I2-Cu1-N2	108.9 (1)
I1-Cu1-N1	117.9 (1)	N1-Cu1-N2	80.6 (1)
I1-Cu1-N2	112.0(1)	Cu1-I1-Cu2	59.78 (2)
I2-Cu1-N1	112.1 (1)	Cu1-I2-Cu2	60.26 (2)

The phenanthroline and acetonitrile H atoms were placed in calculated positions with C—H distances in the range 0.93–0.96 Å and treated as riding atoms with isotropic displacement parameters of  $1.2 U_{\rm eq}({\rm phen})$  or  $1.5 U_{\rm eq}({\rm Me})$  of the parent C atom. The largest peak and deepest hole in the final difference Fourier map were 0.935 and 0.745 Å from atom I2, respectively.

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97*.

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